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Kev indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.058 wR factor = 0.168 Data-to-parameter ratio = 16.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3-(4-Chlorophenyl)-5-methyl-3H-1,3,4oxadiazol-2-one

The molecule of the title compound, C₉H₇ClN₂O₂, is planar. A weak intramolecular $C-H \cdots O$ interaction is observed in the solid state.

Comment

The title compound, (I), and its derivatives were obtained by a facile one-pot ring conversion of 3-arylsydnones in approximately 80% yield. The synthesis of a few related compounds which have been reported earlier have been obtained with difficulty in approximately 30% yield along with isomers from phenylhydrazine (Kametani et al., 1970). Compound (I) and its derivatives can be used as synthetic precursors for other heterocycles. The title compound shows growth inhibition only against P. pyocyanous equal to that of the standard drug norfloxacin (Mallur & Badami, 2000). The present study was undertaken to determine the crystal and molecular structure of (I).



A view of the molecule is shown in Fig. 1. The bond lengths and angles in the oxadiazole moiety are comparable with those of related structures (Du et al., 2004; Öztürk et al., 2004). The dihedral angle between the mean planes of the benzene and oxadiazole rings is $3.63 (8)^{\circ}$. In the crystalline state, a weak intramolecular C-H···O interaction is observed between atoms C7 and O2 (Table 1), forming an S(6) motif (Bernstein et al., 1995).

Experimental

3-(4-Chlorophenyl)sydnone (1 g) was suspended in acetic anhydride (5 ml) at 273 K and an ice-cold solution of bromine (0.5 ml) in acetic anhydride (5 ml) was added with stirring and cooling. 4-Bromo-



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Figure 1 View of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.



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0964 S. Thamotharan et al. • C₉H₇ClN₂O₂ sydnone began to separate. The reaction mixture was then heated on a water bath for 30 min, gradually increasing the temperature to 323– 333 K. Vigorous evolution of CO_2 was observed. The solution was then diluted with water and the resulting solid was filtered off and washed with water and crystallized from absolute ethanol (m.p. 372– 373 K).

Crystal data

 $\begin{array}{l} C_9H_7ClN_2O_2\\ M_r = 210.62\\ Orthorhombic, Pnaa\\ a = 6.8548 (14) Å\\ b = 12.042 (2) Å\\ c = 22.762 (2) Å\\ V = 1878.9 (7) Å^3\\ Z = 8\\ D_x = 1.489 \ {\rm Mg \ m^{-3}} \end{array}$

Data collection

Stoe IPDS-I diffractometer ω scans Absorption correction: by integration (*X-RED*1.22 in *IPDS Software Package*; Stoe & Cie, 1997) $T_{min} = 0.850, T_{max} = 0.914$ 15537 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.058$ $wR(F^2) = 0.168$ S = 1.082140 reflections 128 parameters H-atom parameters constrained Mo $K\alpha$ radiation Cell parameters from 6328 reflections $\theta = 3.5-27.9^{\circ}$ $\mu = 0.38 \text{ mm}^{-1}$ T = 293 (2) KPrism, colourless $0.30 \times 0.20 \times 0.12 \text{ mm}$

2140 independent reflections 1841 reflections with $I > 2\sigma(I)$ $R_{int} = 0.043$ $\theta_{max} = 28.0^{\circ}$ $h = -8 \rightarrow 8$ $k = -15 \rightarrow 15$ $l = -30 \rightarrow 29$

$$\begin{split} &w = 1/[\sigma^2(F_o^2) + (0.0844P)^2 \\ &+ 0.5325P] \\ &where \ P = (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\rm max} < 0.001 \\ \Delta\rho_{\rm max} = 0.29 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta\rho_{\rm min} = -0.19 \ {\rm e} \ {\rm \AA}^{-3} \end{split}$$

Table 1

C-H···O interaction (Å, °).

$D-\mathrm{H}\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
С7−Н7…О2	0.93	2.38	2.994 (3)	124

The methyl H atoms were constrained to an ideal geometry (C–H = 0.96 Å), with $U_{iso}(H) = 1.5U_{eq}(C)$. All remaining H atoms were placed in geometrically idealized positions (C–H = 0.93 Å) and constrained to ride on their parent atoms, with $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: *IPDS Software Package* (Stoe & Cie, 1997); cell refinement: *IPDS Software Package*; data reduction: *IPDS Software Package*; program(s) used to solve structure: *SIR*92 (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEP*-3 (Farrugia, 1997); software used to prepare material for publication: *SHELXL*97 and *PLATON* (Spek, 2003).

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